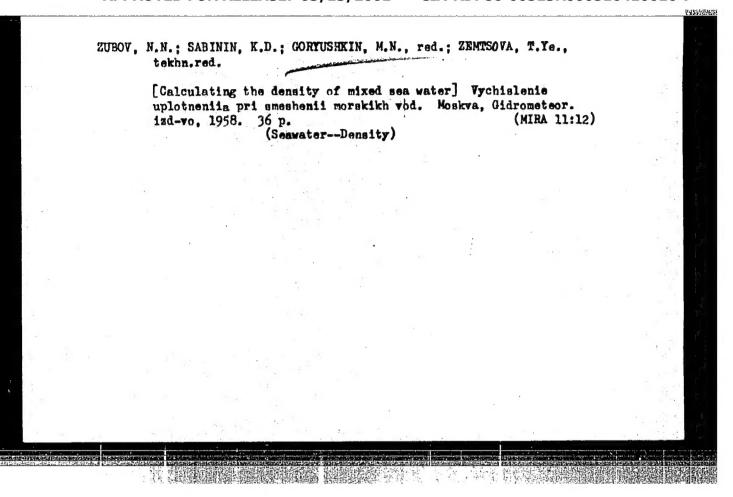
KOCHERGIN, S.V.; GORYUSHKIN, F.F., dorozhnyy master; BORISENKO, D.G., brigadir; GRINEVICHUS, E.A. [Grinevicus, E.]; KURS, V.G., brigadir; SELIONOV, S.I.; BEN°KOVSKIY, V.Ya.; PIRIYEV, A.M.

Letters to the editor. Put1 i put.khoz. 7 no.2:36-37 163. (MIRA 16:2)

1. Zamestitel; nachalinika Rossoshanskoy distantsii Yugo-Vostochnoy dorogi (for Kochergin). 2. Stantsiya Kudinovo, Moskovskoy dorogi (for Goryushkin). 3. Stantsiya Rzhanitsa, Moskovskoy dorogi (for Borisenko). 4. Starshiy dorozhnyy master, stantsiya Klaypeda, Litovskoy dorogi (for Grinevichus). 5. Stantsiya Cherenkhovo, Vostochno-Sibirskoy dorogi (for Kurs). 6. Zamestitel; nachalinika distantsii, Manzovka, Dalinevostochnoy dorogi (for Selionov). 7. Nachalinik otdela zashchitnykh lesonasazhdeniy sluzhby puti, g.Kuybyshev (for Benikovskiy). 8. Zamestitel; nachalinika distantsii, Khachmas, Azerbaydzhanskoy dorogi (for Piriyev).

GORYUSHKIN, Leonid Mikhaylovich; RAZGON, I.M., doktor ist. nauk, prof., otv. red.; NAZARYANTS, T.M., red.; YEFREHOVA, G.A., tekhn. red.

[Social and economic prerequisites of socialist revolution in the Siberian village; the level and characteristics of the development of capitalism in the agriculture of Western Siberia at the end of the 19th and the beginning of the 20th century] Sotsial'no-ekonomicheskie predposylki sotsialisticheskoi revoliutsii v Sibirskoi derevne; ob urovne i osobennostiakh razvitiia kapitalizma v sel'skom khoziaistve Zapadnoi Sibiri v kontse XIX-nachale XX vv. No-vosibirsk, Izd-vo Sibirskogo otd-niia AN SSSR, 1962. 128 p. (MIRA 16:5)

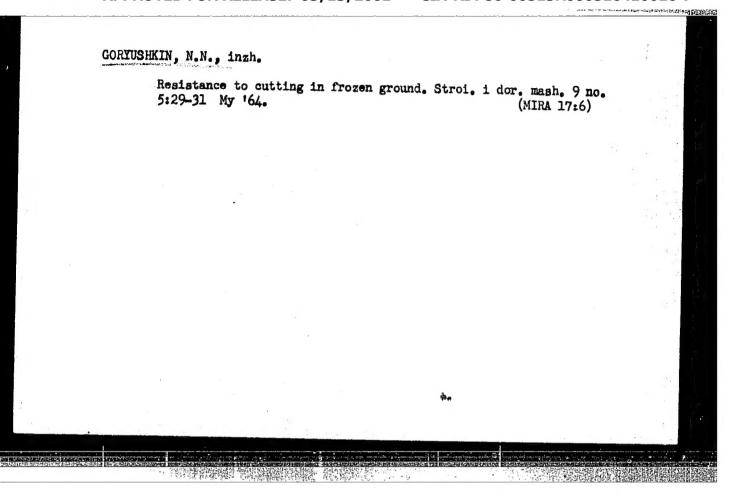


SAMOYLENKO, V.S.; BAGROV, N.A., kend.fig.-matem.nauk, red.; GORYUSHKIN,
M.M., red.; ZEMTSOVA, T.Ye., tekhn.red.

[Formation of the temperature regimen in seas] Formirovanie
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Gidrometeor.izd-vo, 1959. 144 p.

(Quean temperature)

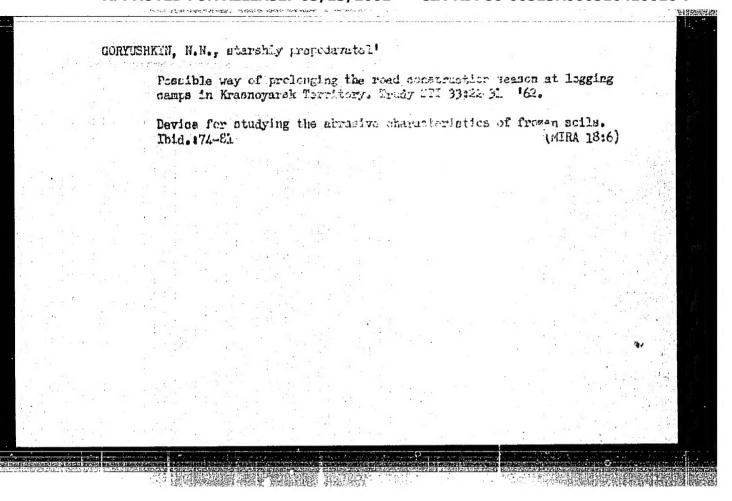
(Quean temperature)



GORYUSHKIN, N.N., starshiy prepodavatel'

Wear of steel as a function of the adherence of abrasive mineral particles to frozen ground. Trudy STI 37:3-14 '64.

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KAUFMAN, D.I.; CORYUSHKIN, V.D.

Automatic line for machining wooden bars. Edul. tekh.-ekon. inform. Cos. nauch.-issl. inst. nauch. i tekh. inform. 17 no.6:59-60 Je 164.

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SHOR, D.I.; BARANOV, V.V.; GORYUSHKIN, V.N.; LEV, M.A.

Basic parameters for sectional reinforced-concrete linings in the horizontal underground mining by the shield method. Trudy TSNIIPodzemshakhtstroia no.3:144-158 '64. (MIRA 18:9)

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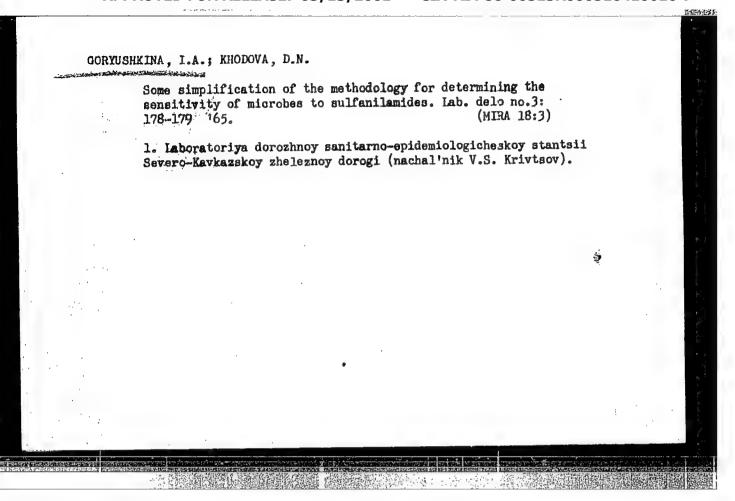
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PANKOVA, R.A.; KHODOVA, D.N.; GORYUSHKINA, I.A.

Study on the survival of dysentery microbes in dryed feces in transportation under the conditions of Northern Caucasus. Zhur. mikrobiol.; epid. i immun. 41 no.6:133-134 Je 164.

(MIRA 18:1)

1. Dorozhnaya sanitarno-epidemiologicheskaya stantsiya Severo-Kavkazskoy zheleznoy dorogi.



L 15757-63 EPR/EWP(j)/EPF(c)/EWT(m)/HDS RM/WW ACCESSION MR: AP5005858 8/0051/63/015/002/0286/0287 AUTHOR: Voloshin, V. A.: Corvushko, A. G.; Kul'chitskiy, V. A. TITLE: Spectroscopic investigation of poly(methyl methacrylate eurorium benzoylacetonate SOURCE: Optika i spektroskopiya, v. 15, no. 2, 1963, 286-287 TOPIC TAGS: rare earth chelate, rare earth, chelate, europium benzoylacetonate, polymer, poly(methyl methacrylate), laser, laser material, absorption spectrum, emission spectrum, europium, benzoylacetome, complex ABSTRACT: A study has been made of the spectroscopic properties of a rareearth chelate in a polymer because rare-earth chelates in the crystalline state are not likely to be used in laser systems owing to too high a concentration of both absorption and emission centers. Two percent of europium 1-phenyl-1,3-butanedione complex (I) activating poly(methyl methacrylate) and crystalline I were both used in the experiment. The main results are given in Fig. 1 of the Enclosure. The absorption spectrum at room temperature $(\lambda_{max} = 308 \text{ m}\mu_s)$ $\epsilon = 1.6 \times 10^{\circ}$) is characterized by a wide region of singlet-singlet transition

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of I. In general, the emission spectra of I are identical in the crystalline

L 15767-63 ACCESSION NR: AP3005858

state and in the polymer and are characterized by the following data: 1) Enission occurs only from the 5Do level except for the very weak, but sharp, 5356-A line which appears in crystalline I at 100K and a 60-min exposure and may be due to a con ment of the $5D_1 \longrightarrow 7F_1$ transition. 2) The $5D_0 \longrightarrow 7F_0$ transition is vary strong, nearly identical in intensity to the 5Do -> 7F1 transition. This indicates the low symmetry of the field in which the europium ion is located. 3) The complete removal of degeneracy from the 5Do -- 7F1 and 500 --- 7F2 transitions also indicates a low-degree of symmetry. 4) The presence of transitions for which AJ exceeds 2 indicates that the europium ion is not at the center of symmetry. 5) The most intense lines occur in the 5Do ---- 7F2 transition. 6) Only four of the seven components of the 5Do --transition were observed, which was apparently due to a loss of film sensitivity. These data prove that both the strength and the symmetry of the field in which the curopium ion is located are identical in the crystalline state and in the polymer. However, there are certain differences in the spectra of the two samples: 1) In the polymer the half-width of lines is 2 to 3 times as great (10-30 Å versus 3-10 Å). 2) The most intense component (5879 Å) of the 5Do --- TI transition in the crystal is the least intense in the polymer. 3) A number of weak lines, 6153, 6218, 6299, and 6328, can evidently be attributed to the superposition of vibration frequency on electron transitions. art. has: 1 figure.

L 61672-65 EXT ACCESSION NR: A	1)/ENT(m)/ENF(t)/ENP(b) Pi-4 1 P5011115 1, V. A.; Goryushko, A. G.; Daviden A.	JP(c) JD/JC UR/0051/65/018/004/0628/063 535.37 ko, N. K.; Klimusheva, G. V.;	9
Eul'chitekiy, V.	A.		
acetonate. I.	ution of the luminescence enectrum uminescence from two resonant leve		
SOURCE: Optika	i spektroskopiya, v. 18, no. 4, 196), 4mg - y-	
ABSTRACT: The and the survey of the protograph of the last photograph of the last photograp	opium compound, luminescence spectruthors obtained and analyzed in det pium benzoylacetonate in the 500-raphed with a spectrograph (STE-1) hic material, which displayed hither minescence spectra of two benzoylas, were identical, except for different lines are tabulated and the winescence are tabulated and tabulated and tabulated and tabulated and tabulated and tabulated are tabulated and tabulated and tabulated and tabulated and tabulated are tabulated and tabulated and tabulated and tabulated and tabulated are tabulated and tabulated and tabulated and tabulated are tabulated and tabulated are tabulated and tabulated and tabulated and tabulated are tabulated are tabulated are tabulated and tabulated are tabulated ar	ail the luminescence spectra of 700 mm band at 20K. The lumine of high resolution and more seem to unnotice details in the acetonate compounds, prepared basion of some lines. The frequencies transitions responsible	neu-
for the lines ar	rved lines are tabulated and the wood identified. The number of lines	IN the shaction is rendar and	1
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	L 61672-65 ACCESION MR: AP5011115 expected when account is taken of the maximum splitting of the levels and of the overlap of transition from the two excited electronic levels 500 and 501 to the 77n levels, and that almost each band of the spectrum is doublet. The doublet nature can be attributed to the presence of two luminescence centers, the details of which will be dealt with in the second part of the article. Orig. ar: has: 2 figures										
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ACCESSION NR: AP4017397

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AUTHOR: Voloshy*n, V. A.; Goryushko, G. G.; Kulichy*tsiky*y, V. O.

TITLE: Energy States of Benzoylacetate rare-earth complexes in a polymethylmethacrylate host.

SOURCE: Ukrayins'kyty fizytchnyty zhurnal, v. 9, no. 2, 1964, 192-195

TOPIC TAGS: rare-earth chealate, chealate luminescence, organic laser material, luminescence europium benzoylacetate, terbium benzoylacetate, europium chealate, terbium chealate, polymethylmethacrylate chealate host, copolymerization, rate-earth-organic complex

ABSTRACT: Brightly luminescing polycrystalline europium benzoylacetate (EBA) in a polymethylmethacrylate (PMM) host luminescess just as brightly as without the host. The basic characteristics of the luminescence spectrum are unchanged, and likewise for the absorption spectra. These were studied between 2800 and 3700 A. Luminescence spectra were taken at room temperature and at liquid nitrogen, hydrogen and helium temperatures with ISP-51 and STE-1 instruments.

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ACCESSION NR: AP4017397

Absorption spectra were taken at room temperature only on a SF-4 apparatus. Terbium benzoylacetate (TRA), in contrast to EBA, ceases luminescing in a PMM host, and its absorption spectrum changes sharply. The comparison is shown in Figures 1 and 2 of Enclosure Ol. The conclusion is that EBA dissolves in PMM, while TBA copolymerizes. "The authors, in conclusion, consider it their pleasant duty to thank their colleagues at the Institute of Physics of the Ukrainian Academy of Sciences, D. F. Sheka and G. V. Kly*musheviy for their assistance and helpful discussions." Orig. art. has 2 figures

ASSOCIATION: Fizy*kotekhniohny*y insty-tut, AN URSR, Kharkov (Physico-Technical Institute, AN URSR)

SUBMITTED: 22Jul63

DATE ACQ: 19Mar64

ENCL: 02

SUB CODE: PH

NO REF SOV: 001

OTHER: 002

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GORYUSHKO, V.Ya. [Horiushko, V.IE.]; MIKHAYLOV, V.F. [Mykhailov, V.F.]; POTRASHKOV, V.I., kand. tekin. nauk; TKACH, G.A. [Tkach, H.A.], kand. tekhn. nauk

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GORYUSHKO. V. Ye. [Horlushko, V.IE.]; TRACHUK, S.V., kand. tekhn. nauk;

MEKINYAN, Yu.G. [Mekintan, IU.H.]

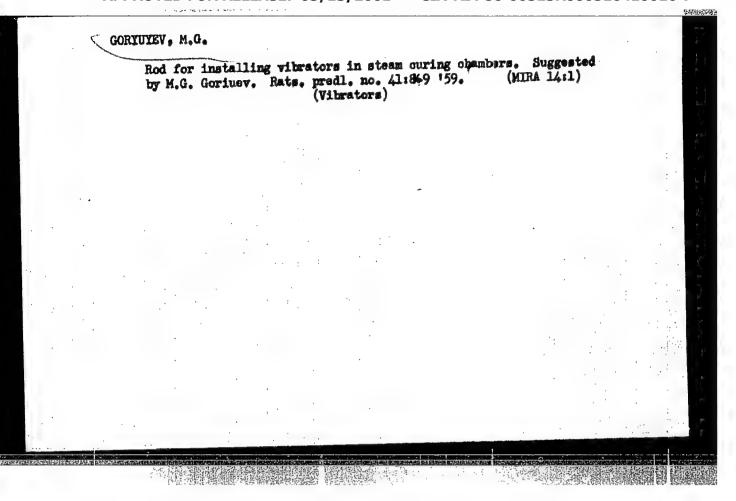
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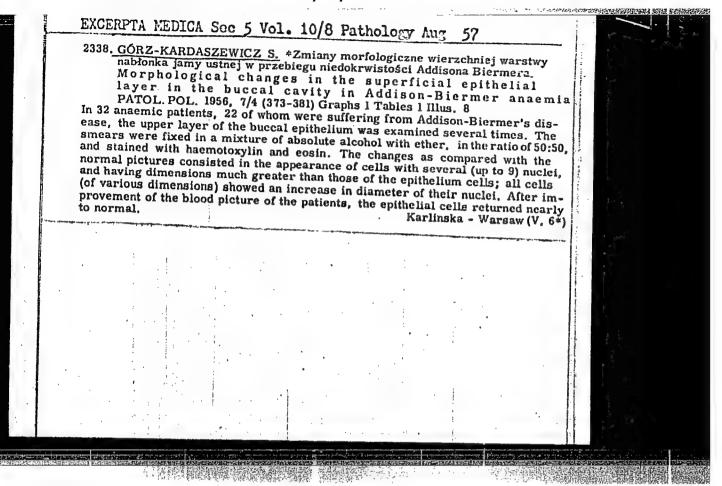


CORZ, A.

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1. Polska Akademia Nauk, Biuro Prezydialme.

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Recent resolutions of the Scientific Secretariat, Polish Academy of Sciences concerning the training of educational personnel. Nauka Polska 9 no.3:239-248 '61.

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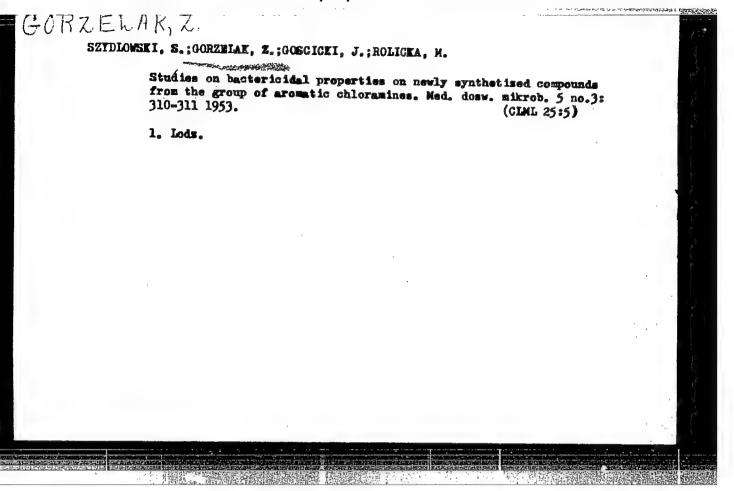
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(BLOOD SEDIMENTATION.

*relation to viscosity of hyaluronic acid)
(HYALURONIC ACID,

*viscosity, relation to blood sedimentation)

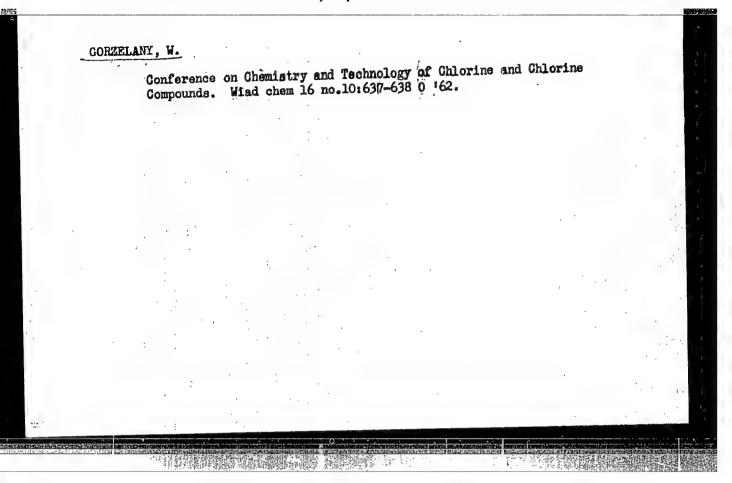
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(Domestic animals) (Chlorodimethylaminopropylphenathiasine)

MATOUSHEK, Iozef [Matousek, Josef]; CHUTA, Ya. [Cuta, J.] tekhnicheskiy sotrudnik; GLAZROVA, Z. [Clasrova, Z.], tekhnicheskiy sotrudnik; GORZHAKOVA, I. [Horzakova, I.], tekhnicheskiy sotrudnik; MATOUSHKOVA, V. [Matouskova, V.]; tekhnicheskiy sotrudnik; SHAKHOVA, G. [Sachova, G.], tekhnicheskiy sotrudnik

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So: B-3,080,964

SMOLYARENKO, D.A.; MATYUSHINA, N.V.; KAPLAN, A.S.; GORZHEVSKAYA, A.V.;
Prinimali uchastiye: ULINSKAYA, Ye.I.; BARYSHEVA, I.V.; ROMAS,
F.D.. AVRUTSKAYA, R.F., red.izd-va; ISLENT'YEVA, P.G., tekhn.
red.

[List of specifications in effect for products of ferrous metallurgy] Percenen' deistvuiushchikh tekhnicheskikh uslovii na produktsiiu chernoi metallurgii; po sostoianiiu na l ianvaria 1959 g. Moskva, Gos.nauchno-tekhn.izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1959. 115 p. (MIRA 13:2)

1. Moscow. TSentral'nyy nauchno-issledovatel'skiy institut
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(for Smolyarenko, Matyushina, Kaplan, Gorzhevskaya). 3. Ukrainskiy
nauchno-issledovatel'skiy trubnyy institut (for Ulinskaya). 4. Nauchno-issledovatel'skiy institut metiznoy promyshlennosti (for
Barysheva). 5. Ukrainskiy institut metallov (for Romas).

(Iron--Specifications) (Steel--Specifications)

BALAKINA, I.A.; BOCHKAREVA, A.I.; GORZHEVSKAYA, A.V.; KAPLAN, A.S.; SMOLYARENKO, D.A., kand. tekhn.nauk; TERENT'YEV, Ye.A.; SOTS, G.A.; TREMBITSKIY, Ya.V.; ULINSKAYA, Ye.I.; KHUTURSKAYA, Ye.S., red. izd-va; KLEYNMAN, M.R., tekhn. red.

[Technical specifications in effect on products of ferrous metallurgy; list as of October 1, 1961] Deistvuiushchie tekhnicheskie usloviia na produktsiiu chernoi metallurgii; perechen' po sostoianiiu na 1 oktiabria 1961 g. Moskva, Metallurgizdat, 1962. 141 p. (MIRA 15:5)

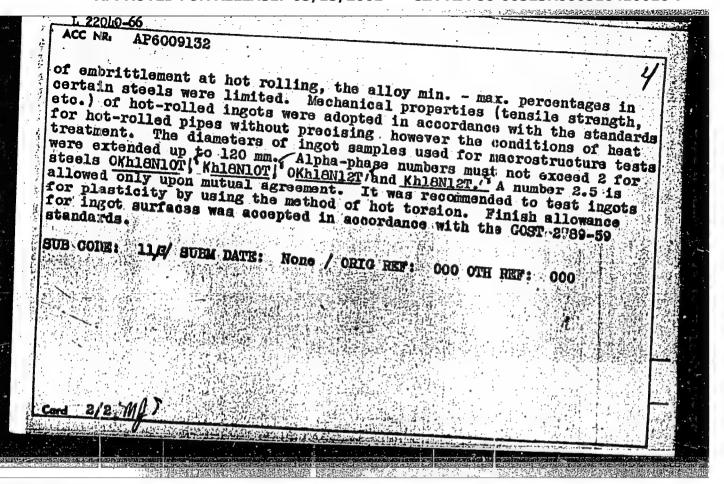
1. Moscow. TSentral'nyy nauchno-issledovatel skiy institut chernoy metallurgii.

(Iron industry-Tables and ready-reckoners)
(Steel industry-Tables and ready-reckoners)

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GORZEVSKAYA, E.G.

SUBJECT

USSR / PHYSICS

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AUTHOR TITLE

GORZEVSKAJA, E.G., PANOVA, N.M.

The Photoproduction of Slow Negative Pions on Complicated Nuclei.

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Issued: 2 / 1957

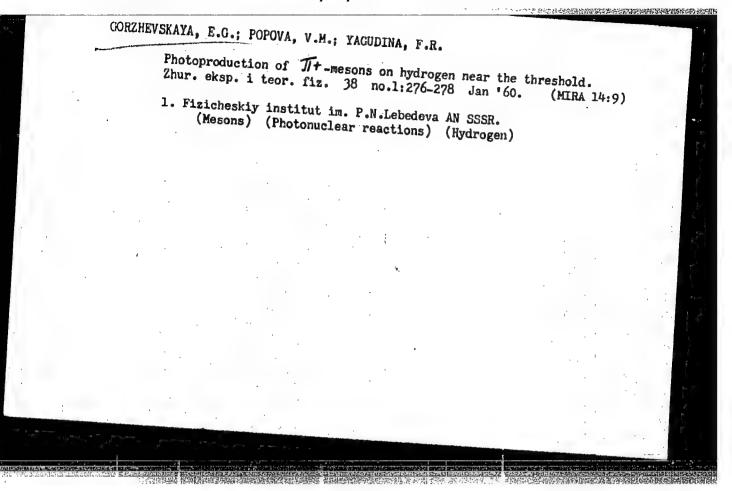
The present work is intended as an investigation of the mechanism of this photoproduction. Above all it is intended to find out whether the photoproduction of mesons in complicated nuclei takes place on a single nucleon or whether this process is more complicated. Furthermore, data concerning the interaction between a slow meson and the trunk of the nucleus are sought. The experiment: NIKFI photoemulsions of the type "P" were irradiated on the synchrotron of the Physical Institute of the Academy of Science by a photon bundle with $E_{max} = 250$ MeV. As the photoenulsions were saturated with heavy water, it was thus possible at the same time to study the photoproduction of negative pions on deuterium. The carrying out of the experiment is discussed in short. Measuring results: On an emulsion surface of 830 cm2,262 mesons, which had come to a standstill in the emulsion, were discovered. Of these 75 and 163 were asigned to the production of negative pions on a light and on a heavy nucleus respectively. In 24 cases the nucleus, on which the meson was produced, could not be identified. The cross section of the production of negative pions with energies of up to 4 MeV on the light and heavy nuclei of the emulsion amounts (after the necessary corrections have been taken into account) to $(2,2\pm0,33).10^{-29}$ cm² and $(8,8\pm0,9).10^{-29}$ cm². A graph and a table illustrate

Dokl. Akad. Nauk 111, fasc. 6,1205-1208 (1956) CARD 2/2 the distribution of stars according to the number of their beams; the traces of mesons and recoil nuclei are not counted on this occasion. A great part of twobeam stars and practically all stars with more than two beams belong to the light nuclei. In many cases only one proton is emitted on the occasion of the

production of a slow negative pion. The angular distribution of these protons in the laboratory system has a marked maximum in the direction of the photon bundle and when photon energy was increased the maximum became even more

Discussion of results: The angular distribution of the protons originating from heavy nuclei and from deuterium (in which negative pions with less than 10 MeV are produced) have the same character, but in the angular distribution of the protons originating from deuterium the maximum is more marked. This difference indicates a considerable influence exercised by the motion of the nucleons in the nucleus. At least in 30% of the cases investigated the photon produces a meson by interaction with one of the nucleons of the nucleus, on which occasion the momentum of the photon is transferred essentially to that nucleon on which the meson is produced. The results found here are a good illustration and proof of the single-nucleon model.

INSTITUTION: Physical Institute "P.N.LEBEDEV" of the Academy of Science in the



ADAMOVICH, M.I.; GONZHEVSKAYA, E.G.; POPOVA, V.M.; YAGUDINA, F.R.

Method for measuring the photoproduction cross section of 7t -mesons on hydrogen near the threshold. Zhur.eksp.i teor.
fiz. 40 no.3:974-976 Mr '61. (MIRA 14:8)

1. Fizicheskiy institut im. P.N.Levedeva Akademii nauk SSSR. (Mesons) (Ionization chamber) (Fhotomuclear reactions)

S/056/61/041/006/023/054 B102/B138

AUTHORS:

Adambvidh, M. I., Gorzhevskaya, E. C., Larionova, V. C.,

Panova, N. M., Popova, V. M., Kharlamov, S. F., Yagudina, F.R.

TITLE:

The energy dependence of the photoproduction cross section of

π⁺ mesons on hydrogen near the threshold

PERIODICAL:

Zhurhal eksperimental noy i teoreticheskoy fiziki, v. 41,

no. 6(12), 1961, 1811-1817

TEXT: The paper gives results of π^+ photoproduction cross section measurements made in the photon energy range from 167 to 212 MeV at an angle $\theta = \arccos(k - 0.93)/kq$, i. e. the angle in the c. m. s. at the contribution of the non-physical region to the dispersion integral vanishes. k denotes the photon momentum, 0.93 is its threshold, q and are momentum and total energy of the pion, θ the angle of emission of the meson; $\theta = c = \mu = 1$. The energy range was chosen so as to satisfy the relation $kn = kq \cos \theta = 0.93$; it holds exactly for 195-MeV photons, for 167 and 212 MeV it is 0.88 and 0.99, which are both close to the threshold value. The photon ray from the synchrotron of the FIAN with a maximum Card $1/\mu \omega$

S/056/61/041/006/023/054 B102/B138

The energy dependence of the ...

energy of 250 Mev was collimated and directed on to the hydrogen target, a brass cylinder of 17 μ wall thickness, placed in a vacuum chamber. The detector was a stack of 50 layers of Mikkin 5%-400 (NIKFI BK-400) emulsion plates. It was placed between two 2cm-thick emulsion blocks and fixed so that the mesons struck its end. The emulsions were evaluated as usual, by Mil-1 (MBI-1) microscopes. All π - μ decay events were selected. An area of 340 cm² yielded 3322 π - μ decays and 64 π decays. The differential photoproduction cross sections were plotted after applying corrections for energy loss, scattering meson decay and background (Fig. 3). The results

photoproduction cross sections were plotted after applying. 3). The results energy loss, scattering meson decay and background (Fig. 3). The results are in good agreement with dispersion theory, where the imaginary part of the resonance amplitude is determined empirically. The experimental results were treated by the method of least squares to find the threshold value of the matrix element of π^+ photoproduction $(-1 d\sigma/d)$ and its dependence on q^2 ;

$$\frac{(q/k)(1+x/M)^{-2}, M - nucleon mass. For 0.17 \langle q^2 / 0.74}{\frac{i}{\chi} \frac{d\sigma}{d\Omega} \left[10^{-29} \frac{cM^3}{cmepad} \right] = (1,90 \pm 0,15) - (0,34 \pm 0,22) q^3,$$
 (5)

Card
$$2/\beta$$
 $\frac{1}{\chi} \frac{d\sigma}{d\Omega} \left[10^{-20} \frac{cM^2}{cmepa\partial} \right] = (2,39 \pm 0,21) - (2,87 \pm 0,93) q^2 + (2,80 \pm 1,0) q^4,$ (6)

S/056/61/041/006/023/054 B102/B138

The energy dependence of the ...

was found. The threshold value was determined from power expansions of the squares of the matrix elements, $a_0 = (1.90 \pm 0.15) \cdot 10^{-29} \text{ cm}^2/\text{steradian}$, which is in good agreement with the theoretical value, $a_0 = 2.04 \cdot 10^{-29} \text{ cm}^2/\text{sterad}$. Experimentally, $\sigma'/\sigma' = 1.34 \pm 0.11$ was found. Using the theoretical a_0 value, the calculated value is $\sigma'/\sigma' = 1.28$. The pion photoproduction cross section as a function of the photoproduction amplitudes is given by $\frac{dc/d\Omega}{dc/d\Omega} = \frac{(g/R)(|F_1|^2 + |F_2|^2 - 2ReF_1^2F_1\cos\theta + \frac{1}{2}\cos\theta}{(9)}$

 $do/d\Omega = (q/k) \{ |F_1|^2 + |F_4|^2 - 2\text{Re } F_1^* F_3 \cos \theta + + \frac{1}{2} \sin^2 \theta \{ |F_3|^2 + |F_4|^2 + 2\text{Re } F_3^* F_3 + 2\text{Re } F_1^* F_4 + 2\text{Re } F_3^* F_4 \cos \theta \} \},$

 $F_{1} = \sqrt{2}F_{10} - \sqrt{2}F_{11}\cos\theta, \qquad F_{2} = \sqrt{2}F_{20},$ $F_{3} = \sqrt{2}F_{30} + \sqrt{2}F_{31}/(1 - \beta\cos\theta), \qquad F_{4} = \sqrt{2}F_{41}/(1 - \beta\cos\theta);$

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5/056/61/041/006/023/054 B102/B138

The energy dependence of the

 β denotes pion velocity. From experimental data for 15 and 165° c. m. s. the amplitudes were calculated for 185-Mev photons:

 $[(F_{10})_1 = (1.81 \pm 0.034) \cdot 10^{-8}, \qquad (F_{11} + F_{20})_1 = -(0.105 \pm 0.034) \cdot 10^{-8}]$ $(F_{10})_3 = -(1.81 \pm 0.034) \cdot 10^{-3}, \qquad (F_{11} + F_{30})_3 = (0.105 \pm 0.034) \cdot 10^{-3}.$

The authors thank Professor P. A. Cherenkov for help, A. M. Baldin and A. I. Lebedev for discussions and A. A. Svetlov, Engineer, for assistance. There are 5 figures, 2 tables, and 15 references: 3 Soviet and 12 non-Soviet. The four most recent references to English-language publications read as follows: J. Hamilton, W. S. Woolcock. Phys. Rev. 118, 291, 1960; N. P. Samios. Phys. Rev. Lett., 4, 470, 1960; M. Derrick et al. Phys. Rev. Lett., 5, 230, 1960; A. F. Dunaitsev et al. Proc. 1960 Ann. Intern. conf. on high energy physics at Rochester, Publ. Univ. Rochester 1961, p. 181.

Fizicheskiy institut im. P. N. Lebedeva Akademii nauk SSSR ASSOCIATION:

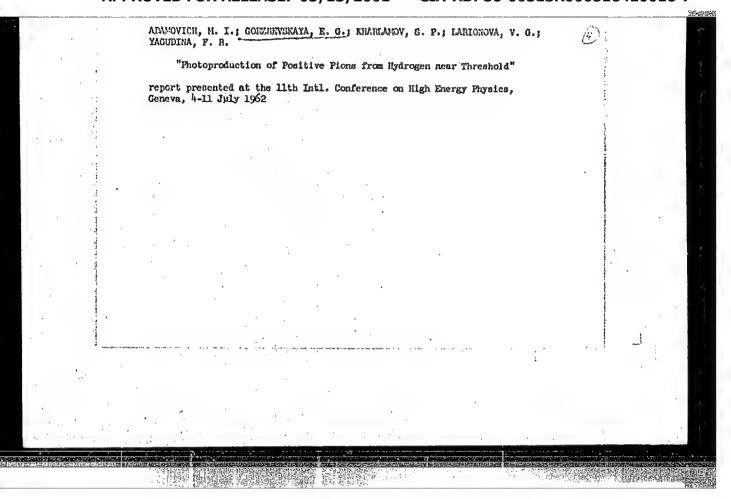
(Physics Institute imeni P. N. Lebedev of the Academy of

Sciences USSR)

SUBMITTED:

July 31, 1961

Card 4/#



s/056/62/043/003/057/063 B104/B102

AUTHORS:

Adamovich, M. I., Gorzhevskaya, E. G., Yagudina, F. R.

TITLE:

The production of π -photomesons at angles of 25-36° in the

energy range 152-162 Mev

PERIODICAL:

Zhurnal eksperimental noy i teoreticheskoy fiziki, v. 43,

no. 3(9), 1962, 1113-1116

TEXT: This study was directed to establishing the differential photoproduction cross section of π^+ -mesons when the momentum transfer kw-kq cos θ is close to its threshold value of 0.93, k and q being respectively the momenta of photon and pion, and w the total ion energy in the c.m.s. The mesons emitted by a thin polyethylene film at an angle of about 300 from the photon beam were examined by a method described previously (M. I. Adamovich et al., ZhETF, 40, 974, 1961). Zymax was 175 Mev. All π - μ decay events were recorded. The ends of the pion and muon traces were established for checking. The results (Table) are in good agreement with the calculations. The threshold value of $(1/\chi) d\sigma/d\Omega$ is $(2.18 \pm 0.37) \cdot 10^{-29}$ cm²/sterad. The threshold value calculated from Card 1/2

The production of π^+ -photomesons at...

s/056/62/043/003/057/063 B104/B102

Panov's formula is 1.99·10 $^{-29}$ cm $^2/sterad$. The mean values of $d\sigma/d\Omega$ and $(1/\chi)d\sigma/d\Omega$ agree well with the values for $k\omega$ - kq cos θ = 0.93 as extrapolated from experimental data. There are 2 figures and 1 table.

ASSOCIATION: Fizicheskiy institut im. P. N. Lebedeva Akademii nauk SSSR

(Physics Institute imeni P. N. Lebedev of the Academy of

Sciences USSR)

SUBMITTED:

July 3, 1962

Table. Results of measure-

Legend: E_{γ} - mean photon energy, laboratory system; E_{π} - mean pion energy in the energy interval of the photons; M - proton mass;

 $\chi = (q/k)(1+\omega/M)^2.$

E _γ , MeV	E _n , MeV	q*	ku — kq cos 9	da'dQ, 10° cm//cmepad	X dΩ + 10° см³/стерад
153,4 155,7 157,6 159,3 160,8	3,8 7,3 9,7 11,6 13.3	0,023 0,048 0,069 0,088 0,104	0,84 0,83	0,32±0,054 0,39±0,070 0,43±0,077 0,40±0,076 0,39±0,097	2,70±0,46 2,26±0,41 2,12±0,38 1,77±0,34

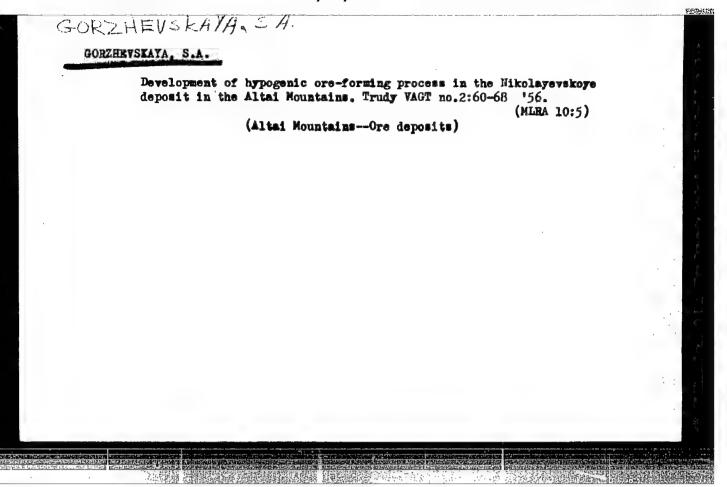
Card 2/2

Varieties or satisfements. Min. shor. no.5:324-329 '51. (MLPA 9:12)

1. Veccoyumnyy institut mineral'nego syr'ya, Moskva.
(Smithsonite)

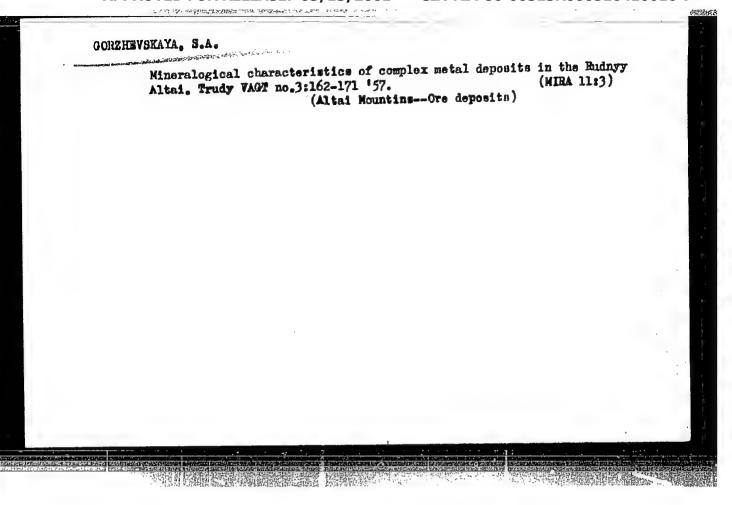
Dissertation: "Application of Methods of Descriptive Mineralogy to the Investigation of Oxidized Ore, For Example, Chrisin Polymetallic Demosits," Cand Geol-Kin Sci, Inst of Geological Sci, Acad Sci USSR, F May 54. (Vechernyaya Moskva, Moscow, 29 Apr 54)

SO: SUM 243, 19 Oct 1954



"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516410016-7



"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516410016-7

GORZHEVSKAYA, S. A. YORZheusiraya, S. A.

"Element-Impurities in Polymetal Deposits of the Rudnyy Altai";

report delivered in the Petrographic Section, 4 April to 7 June 1957.

Chronicle of the Activity of the Petrography Section, Byulleten; Moskovskogo Obshchestva Ispytateley Prirody, Otdel Geologicheskiy, 1957, No. 6, pp. 118-122, 1957.

GOTZHEVSKAYA S.A. 5-6-29/42 Gorzhevskaya S. A. AUTHOR: Element -Impurities in Polymetal Deposits of the Rudnyy TITLE: Altay (Elementy primesi v polimetallicheskikh mestoroshdenivakh Rudnogo Altaya) Byulleten' Moskovskogo Obshchestva Ispytateley Prirody, PERIODICAL: Otdel Geologicheskiy, 1957, # 6, pp 142-143 (USSR) 1.L.32 Altay ores are typical polymetal deposits of low-tem-ABSTRACT: perature stage which originated underhypabyssal conditions. This conclusion is confirmed by the presence in the Altay ores of Te, Se and Hg. The investigations carried out as to distribution of element-impurities in the single-mineral fractions and ores of the Altay deposits indicate that many rare and scattered elements of the Altay polymetal ores may be of practical importance. These are As, Sb, Cd, Te and Se, and possibly also In, Tl, Ge and Hg. At the present time, the data available are not sufficient to determine the industrial value of these elements. AVAILABLE: Library of Congress Card 1/1

AUTHORS:

Ginzburg, A. I., Gorzhevskaya, S. A. Yerofeyeva, Ye. A., Sidorenko, G. A.

SOV/7-58-5-10/15

TITLE:

On the Chemical Composition of the Cubic Titanium-Tantalum Niobates (O khimicheskom sostave kubicheskikh titano-tantalo-niobatov)

uropato.

PERIODICAL:

Geokhimiya, 1958, Nr 5, pp 486 - 500 (USSR)

ABSTRACT:

The specific properties of the so-called mineral group are described in detail in the beginning; then the division into the perovskite type (ABX₃) and pyrochlorine type (A₂B₂X₇) is discussed. 22 chemical and x-ray analyses (Table 3) are the basis of this paper. A number of analyses are plotted in several diagrams of ternary systems:Nb - Ti - Ta (Fig 1); A - B - X (Fig 5); Nb - Ti, Zr - Ta (Fig 6); Ca - TR - U - Th (Fig 7). The dependence of the lattice constant on the TiO₂ content in the perovskite group (Fig 2) and in the pyrochlorine group (Fig 3) is also shown. The result of the paper is a classification of the mineral groups investigated (Table 2). The empiric formulae of minerals greatly differ from the

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On the Chemical Composition of the Cubic Titanium-Tantalum Niobates

SOY/7-58-5-10/15

theoretical formulae generally adopted for them. A deficiency of cations in the group "A" was found. In connection herewith the formula A_{n-x} $B_{p}^{T}X_{q}$ is proposed where x denotes the value determining the deficiency in the atomic numbers of the group "A". For the pyrochloric type the formula then reads $A_{2-1}B_2X_7$, and for the perovskite type A1-xBX3, or A2-xB2X6. The atomic proportion of the cations of the group "A" in the cubic titaniumtantalum niobates ranges from 2,0 to 0,5, a definite dependence between the extent of the cation dericiency in the group "A" and the content of titanium, zirconium, uranium, thorium and water in minerals having been observed. The usual minerals with an increased cation deficiency in the group "A" are metamictic minerals. There are 9 figures, 3 tables, and 23 references, 15 of which are Soviet.

ASSOCIATION:

Vsesoyuznyy institut mineral'nogo syr'ya, Moskva (All Union

Institute for Mineral Raw Materials, Moscow)

SUBMITTED: Card 2/3

March 17, 1958

APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516410016-7"

On the Chemical Composition of the Cubic Titanium-Tantalum Niobates

SOV/7-58-5-10/15

Card 3/3

GINZBURG, A.I.; GORZHEVSKAYA, S.A.; YEROFEYEVA, Ye.A.; SIDORENKO, G.A. Chemical composition of tetragonal titanium-tantalum-niobates. Geokhimiia no.1:11-30 60. (MIRA 13:6) (Fergusonite)

s/081/62/000/010/028/085 B177/B144

AUTHORS :

Ginzburg, A. I., Gorzhevskaya, S. A.

TITLE:

Characteristics of titanium-tantalum-niobates

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 10, 1962, 108 - 109, ab-

stract 10361 (Sb. "Geol. mestorozhd. redk. elementov", no. 10,

M., Gosgeoltekhizdat, 1960, 5 - 10) 5 mg "

TEXT: The composition of titanium-tantalum-niobates is conventionally represented by the formula AmBpXq, where A and B combine cations of closely similar dimensions. In many of them, the group A cations are less. strongly bound with oxygen than group B cations. Group A includes cations with large R: Ca, Na, Y, TR, Th, U, and to a lesser extent K, Pb, Ba, Sr, Mn and others. Group B includes cations having relatively small R; Ti, Nb, Ta, W and also Al, Si, P and others. For several titaniumtantalum-niobates, group & may usefully be subdivided into two sub-groups, A and A2, whereof A includes Ca, Na, U, Th, TR whilst A2 includes Fe2+

Card 1/2

S/081/62/000/010/028/085 B177/B144

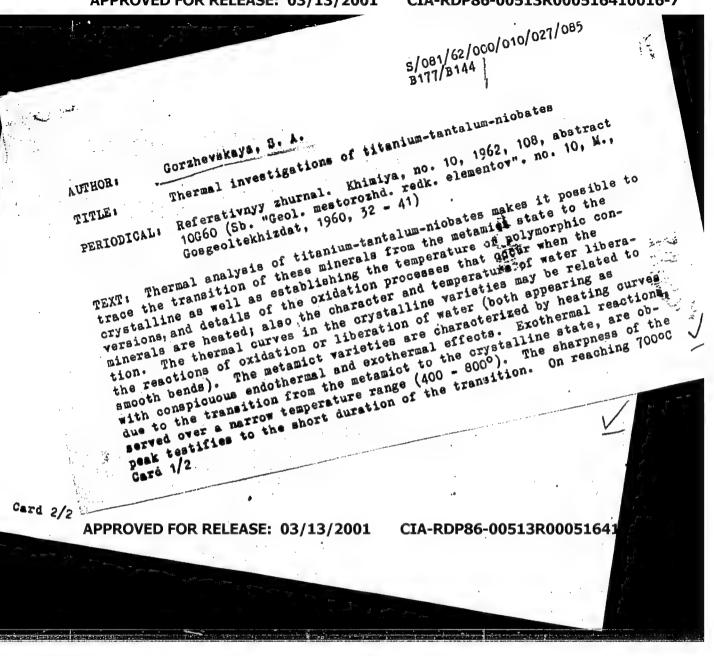
Characteristics of ...

Mn, Mg. In many cases a deficiency of cations exists in group A. In the minerals described, both isovalent and heterovalent substitutions are widely developed. It is observed that, as the ratio between atomic quantities of cations in group A and the atomic quantities of oxygen increases, the syngony of titanium-tantalum-niobates decreases. The variable composition of many minerals is governed by the different processes that change them, which mostly take place under hydrothermal conditions. Probably the processes of change are connected with the extraction of water and with the partial leaching of cations in group A besides others. [Abstracter's note: Complete translation.]

Card 2/2

"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000516410016-7



S/081/62/000/010/026/085 B177/10144

AUTHOR

Gorzhevskaya, S. A.

TITLE:

The chemical composition of titanium-tantalum niobates

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 108, abstract 10G59 (Sb. "Geol. mestorozhd. redk. elementov", no. 10, M., Gosgeoltekhizdat, 1960, 41 - 47)

TEXT: Results of conversions from collected chemical analyses are shown plotted in triangular diagrams to represent the interrelations between atomic quantities of groups A and B. These confirm the presence of a continuous isomorphic series between the Nb and Ta varieties of the same minerals and also the existence of a wide isomorphism between Nb and Ta, but continuous transitions from the Ti minerals to Ta are not established. A much close kinship is disclosed between Ta and Nb than detween Ta and Ti. Minerals composed substantially of Ca are characterized by a relatively constant chemical composition. They are distinguished by a slight content of TR and by the appearance also of TR as the Ca content diminishes. Atomic quantities as high as 25% of Ca, and 20% atomic values

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CIA-RDP86-00513R000516410016-7

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of TR, are noted in the U varieties. All titanium-tantalum-niobates subdivide into several groups, according to the relationship of Ti and TR.

[Abstracter's note: Complete translation.]

S/081/62/000/010/034/085 B177/B144

AUTHORS: Sidorenko, G. A., Gorzhevskaya, S. A.

TITLE: Cubic titanium-tantalum-niobates. X-ray analysis.

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 10, 1962, 110, abstract 10G69 (Sb. "Geol. mestorozhd. redk. elementov", no. 10, M., Gosgeoltekhizdat, 1960, 64 - 71)

TEXT: Cubic titanium-tantalum-niobates crystallize in two structural types: perovskite and pyrochlore. Many minerals in the perovskite group are pseudo-cubic with a 7.64 - 7.68 kX. It is found that a regularly increases with increasing Nb content and with decreasing Di content. This provides a possibility of determining the Nb content from the value of a. In the effect of the dimensions of groups B and A cations on the dimensions of a unit cell is noted in the structural type of pyrochlore. An increased it content causes a reduction of a. In proportion to the substitution Nb Ti in group A, the following isomorphic substitutions occur: Ca and Na are replaced by TR, U and Th. The titanium-uranium-rare earths constitute minerals having elementary cells of reduced dimensions. The

Card 1/2

Cuoic titanium-tantalum-niobates...

S/081/62/000/010/034/065 B177/B144

greatest cell dimensions are those typical of minerals in the tantalum and niobium series, which possess a more constant chemical composition (microlites and pyrochlores). With minerals having variable composition, containing Ti, U and TR, the dimensions of an elementary cell decrease. Minerals having a more constant chemical composition possess a crystalline structure, while those whose composition is complex are metamict. A definite relation exists between the chemical composition, structure and the state. [Abstracter's note: Complete translation.]

Card 2/2

5/081/62/000/010/030/085 B177/B144

AUTHOR:

Gorzhevskaya. S. A.

TITLE:

Cubic titanium-tantalum niobates. Chemical composition

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 109, abstract 10G64 (Sb. "Geol. mestorozhd. redk. elementov", no. 10, M.,

Gosgeoltekhizdat, 1960, 71 - 84)

TEXT: Cubic titanium-tantalum niobates are characterized by a complex and variable chemical composition. The minerals so described differ sharply as regards relationship between atomic quantities of group A and group X ions, with noticeably smaller variations of atomic quantities of group B cations. In minerals of the perovskite type a broad isomorphism is noted between Ti and Nb. with simultaneous replacement of Ca for TR. Na and Th in group A. A typical feature is the almost constant enrichment in Ti. Minerals of the pyrochlore type are characterized by their extremely inconstant chemical composition. Among them, minerals of the microlite group are distinguished by showing the least inconstancy and the lowest content of TR and U. In minerals having a variable composition, the additional constants characterizing the mineral varieties are Card 1/2

s/081/62/000/010/030/085 B177/B144

Cubic titanium-tantalum...

the ratios CaO: (UO₂ + UO₃) and Nb₂O₅: Ta₂O₅. In accordance with these ratios, the mineral priazovite, described by Yurk, should be classed with blomstrandite. Three groups of minerals among the cubic titanium-niobates can be distinguished, according to their chemical composition, as varieties heavily enriched with Nb, Ti - Nb and varieties enriched with Ti. [Abstracter's note: Complete translation.]

Card 2/2

5/081/62/000/010/032/085 B177/B144

AUTHORS:

Ginzburg, A. I., Gorzhevskaya, S. A.

TITLE:

The composition of the Cubic titanium-tantalum niobates.

rare-earth elements

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 10, 1962, 109, abstract

10G66 (Sb. "Geol. mestorozhd. redk. elementoy", no. 10, M.,

Gosgeoltekhizdat, 1960, 84 - 89)

TEXT: The minerals of the perovskite type related to ultrabasic alkaline intrusive complexes, are highly selective Ce minerals, though they also contain La, Nd, and some Sm. Pyrochlores from carbonatites, alkaline beds and their pegmatites possess a constant TR composition. There is a marked predominance in them of elements in the Ce group, with slight quantities of Gd and Dy. In pyrochlores from albitites related to subalkaline granitoids, TR of the Y sub-group occur in slight quantities. In pyrochlore-type minerals encountered in granitic pegmatites, the TR content is very variable. A typical feature of them is the higher content of the middle members of the series TR - Sm, Gd, Dy, and sometimes Ce. Abstracter's note: Complete translation. Card 1/1

CIA-RDP86-00513R000516410016-7" APPROVED FOR RELEASE: 03/13/2001

s/081/62/000/010/031/085 B177/B144

AUTHOR :

Gorzhevskaya, S. A.

TITLE:

Conversions from the chemical analyses of cubic titanium-

tantalum-niobates

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 109, abstract 10G65 (Sb. "Geol. mestorozhd. redk. elemantov", no. 10, k.,

Gosgeoltekhizdat, 1960, 100 - 107)

TEXT: The sum of the atomic quantities of groups A and B cations in cubiq titanium-tantalum-niobates is calculated from the atomic quantities of elements therein. The quantity of cations in group B is known to wary only slightly, whereas that in group A is liable to extreme fluctuations. Atomic quantities of cations in group A vary from 2.0 to 0.5, i.e. the minerals are characterized by a certain deficiency of group-A cations. Minerals containing a considerable quantity of Ti and U are always characterized by a greater deficiency of group-A cations. Minerals with a considerable deficiency of group-A cations are always in a metamict state, whereas those having no such deficiency are usually crystalline.

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Conversions from the chemical ...

With a high deficiency of group-A cations, a high water content is also characteristic. A definite relation exists between this deficiency and the quantity of group X anions: $A_{2-z}B_2X_{7-z}$. A relation also exists between the deficiency and the quantity of water in the mineral (n). Its value is n=z and n=2z. The general formula for titanium-tantalumniobates is $A_{2-z}B_2X_{7-z}$ on $A_{2-z}B_2X_{7-z}$. Abstracter's note: Complete translation.

Card 2/2

s/081/62/000/010/037/085 B177/B144

Sidorenko, G. A., Gorzhevskaya, S. A.

Tetragonal tantalum-niobates. X-ray structural analysis

TITLE:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 110, abstract 10672 (Sb. "Geol. mestorozhd. redk. elementov", no. 10, M., PERIODICAL:

Gosgeoltekhizdat, 1960, 129 - 136)

TEXT: Among all titanium-tantalum-niobates, the minerals which crystallize in tetragonal syngony belong to three structural typesize fergusonite, rutile and tapiolite-mossite. Minerals of the fergusonite group are usually found in the metamict state. Tetragonal syngony of fergusonite (3, 5.15, c, 10.89 kX) has been established in a non-metamict specimen. It is isostructural with scheelite. Heating the mineral causes a reduction of syngony to monoclinic with parameters a = 5.05, b = 10.89, c = 5.27 kX, \(= 85°30'\). The reduction of symmetry is possibly due to the ordering of Y and No ions. A natural monoclinic modification of fergusonite (a 5.12, b 10.89, c 5.20 kX, \(\beta = 88010' \) was recently discovered among granites, in the highest-temperature mineral associations.

Card 1/2

CIA-RDP86-00513R000516410016-7" APPROVED FOR RELEASE: 03/13/2001

S/081/62/000/010/036/085 B177/B144

AUTHOR:

Gorzhevskaya, S. A.

TITLE:

Tetragonal tantalum-niobates: their chemical composition and the conversion of their chemical analyses

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 110, abstract 10G71 (Sb. "Geol. mestorozhd. redk. elementov", no. 10, K., Gosgeoltekhizdat, 1960, 140 - 144)

TEXT: Minerals of the fergusonite structural type belong by their chemical composition to the rare-earth tantalum-niobates having the general formula ABX4. By contrast with the cubic titanium-tantalum-niobates, the relation between group A and B cations and oxygen does not vary appreciably. All these minerals are substantially niobium-bearing; more widespread in nature than the Ta varieties. The Nb varieties are almost continuous ismorphic series between YNbO4 and YTaO4 is noted. From the interrelation of the atomic quantities of Ca, TR, U + Th, it is

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Tetragonal tantalum-niobates ...

established that all tetragonal tantalum-niobates are likewise rare earths, in which the TR elements predominate in group A (usually > 80%). The content of oxides of U and Th is very variable, fluctuating from 0.10 to 9.17% for U, and from traces to 6.8% for Th. Usually U > Th. Conversion of analyses of fergusonites leads to the following formula: (Y, Yb, Dy, Nd)(Nb, Ta, Ti)04. Spectral analysis always reveals the presence of Be and Ge; impurities include Al, Si, Zr, Fe, Mn, Mg and Pb; En is constantly present. [Abstracter's note: Complete translation.]

Card 2/2

S/081/62/000/010/039/085 B177/B144

AUTHORS: Ginzburg,

Ginzburg, A. I., Gorzhevskaya, S. A.

TITLE:

Tetragonal tantalum-niobates. Composition of rare-earth

elements

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 10, 1962, 110, abstract 10G73 (Sb. "Geol. mestorozhd. redk. elementov". no. 10, M.,

Gosgeoltekhizdat, 1960, 144 - 152)

TEXT: Fergusonites are distinguished from other titanium-tantalumniobates by their content of Y and TR Y sub-groups. The content of Y
fluctuates from 40 to 70% of the entire TR content. Different genetic
types of fergusonites are characterized by a specific TR composition. In
some types of deposit, fergusonite is a substantially ytterbium-bearing
mineral (unsubstituted granitic pegmatites); in others, dysprosiumytterbium-bearing (accessory in granites); substantially dysprosiumytterbium-bearing (accessory in granites); substantially dysprosiumbearing (quartz albitites connected with granosienites); cerium-dysprosiumbearing (alkaline pegmatites); neodymium-dysprosium-bearing (albitised
alkaline beds and albitised granitic pegmatites). The ratios in

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fergusonites IY: Ce, Yb: Dy, and Ce: Nd may be regarded as indicative, enabling one to judge the relation of fergusonites to various intrusive beds and to different genetic types of deposits. [Abstracter's note: Complete translation.]

"APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516410016-7

CINZEURG, A.I.; CORZHEVSKAYA, S.A.; YEROFEYEVA, Ye.A.; SIDORENKO, G.A.;

MALYSHEV, I.I., red.; POLYAKOV, M.V., red.; RODIONOV, G.G., red.;

STEPANOV, I.S., red.; TROKHACHEV, P.A., red.; FAGUTOV, V.P., red.;

KHRUSHCHOV, N.A., red.; CHERNOSVITOV, Yu.L., red.; SIMANENKOV, I.V.,

red.; SHCHERBINA, V.V., red.; EYGELES, M.A., red.; NEMANOVA, G.F.,

red.izd-va; HYKOVA, V.V., tekhm.red.

[Titanates, tentalates, and niobates] Titano-tantalo-niobaty.

Moskva. Gos. nauchno-tekhm.izd-vo lit-ry po geol.i okhrane nedr.

Part 1. 1968. 166 p. (Geologiia mestorozhdenii redkikh elementov,

no.10).

(NIRA 14:6)

(Titanates)

(Tantalates)

(Niobates)

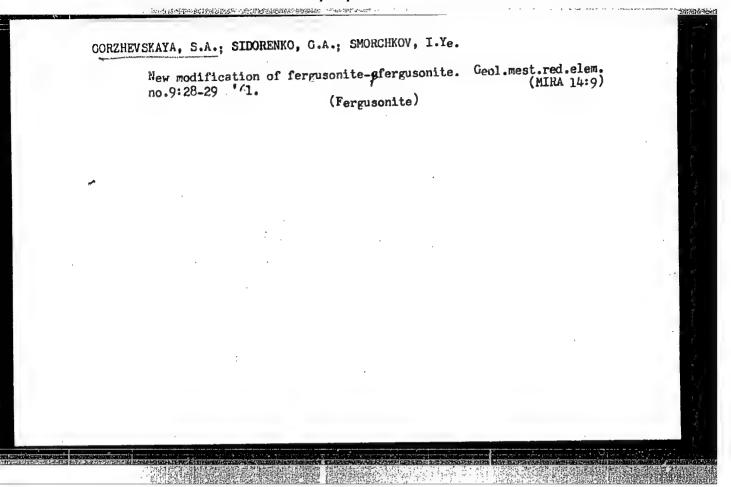
AMBARTSUMYAN, TS.L.; BASALOVA, G.T.; GORZHEVSKAYA, S.A.; NAZARENKO, N.G; KHODZHAYEVA, R.P.; PCHELINTSEVA, G.M., red.; MAZEL', Ye.I., tekbn.

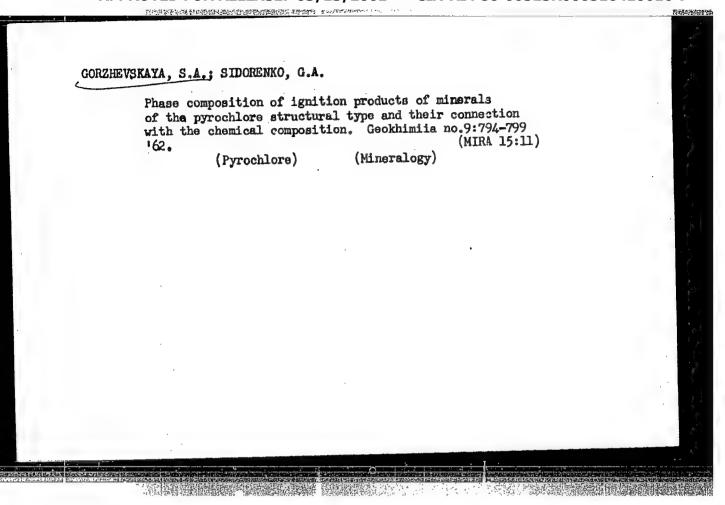
[Thermal investigation of uranium and uranium-containing minerals] Termicheskie issledovanija uranovykh i uransoderzhashchikh mineralov. Moskva, Gos. izd-vo lit-ry v oblasti atomnoi nauki i tekhniki, 1961. 146 p. (MIRA 14:11)

(Uranium-Analysis)

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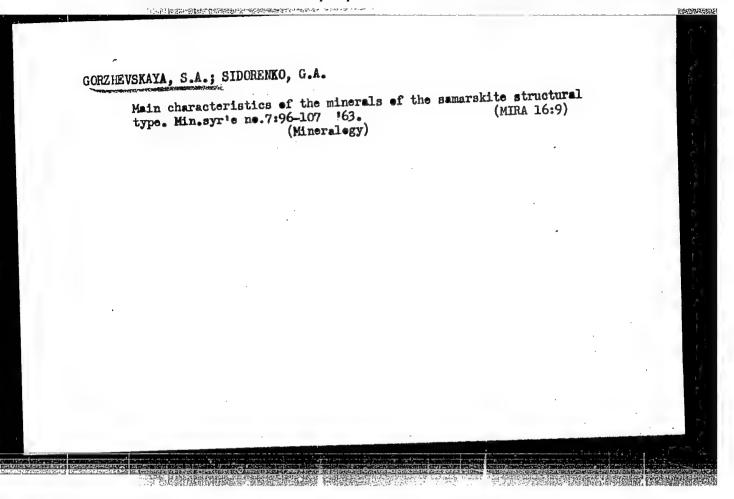


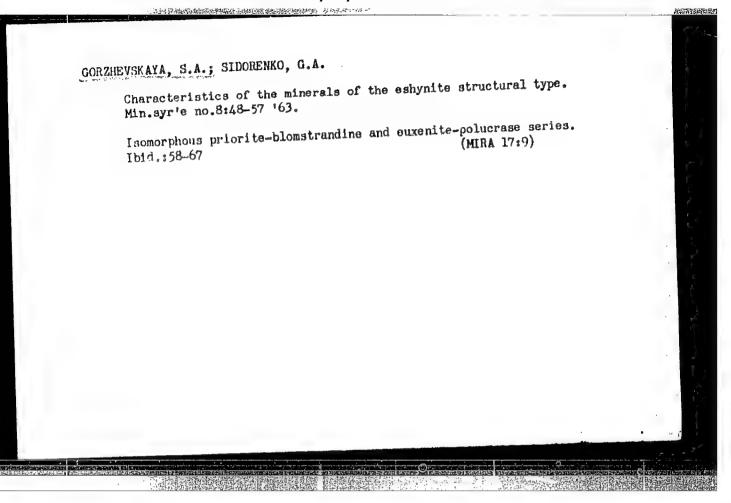


GORZHEVSKAYA, S.A.; SIDORNEKO, G.A.

Find of a crystalline variety of lyndochite. Dokl. AN SSSR 146 ne.5: (MIRA 15:10)

1. Vsesoyuznyy nauchno-issledovatel skiy instutit mineral nogo syr'ya. Predstavleno akademikom D.I.Shcherbakovym. (Lyndochite)





GORZHEVSKAYA, Susanna Aleksandroyna; SIDORENKO, Galina Aleksandrovna;
GINZBURG, A.I., glavnyy red.; POLYAKOV, M.V., zamentitel¹ glavnogo
red.; APEL¹ISIN, P.R., red.; GRIGOR¹EV, V.M., red.; RODIONOV, G.C.,
red.; STEPANOV, I.S., red.; TROKHACHEV, P.A., red.; FAGUTOV, V.P.,
red.; CHERNOSVITOV, Tu.L., red.; SHMANENKOV, I.V., red.; SHCHERBINA,
V.V., red.; EYGELES, M.A., red.

[Titano-tantalo-niobates. Part 2.] Titano-tantalo-niobaty.
Moskva, Nedra. Pt.2. 1964. 115p. (Geologiia mestorozhdenii
redkikh elementov, no.23)

(MIRA 18:1)

CORZHEVSKAYA, S.A.; GREKULOVA, L.A.; SIDORENKO, G.A.

Physical properties and composition of columbite-tantalites. Min.sbor.

18 no.3:257-269 *64.

1. Vsesoyuznyy nauchno-issledovatel*skiy institut mineral*nogo
syr*ya, Moskva.

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GORZHEVSKAYA, S.A.; LUGOVSKOY, G.P.; SIDORENKO, G.A.

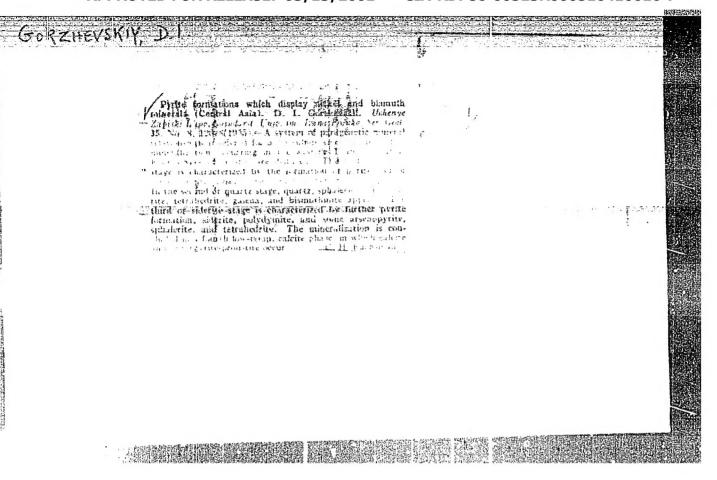
First find of samiresite in the Soviet Union. Dokl. AN SSSR 162
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Submitted March 17, 1964.

Discovery of the rare mineral zinkenite, Min, sbor, no.9:313-314
155.

1.L'vov. Gosudarstvennyy universitet imeni Ivan Franko.
(Zinkenite)

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	Section of the sectio
Gorzheuskij	
user/Geology	[전기기] 발표 원 5.5명에 요면한 제상이는 그는 그는 요리는 이번 병이나도 걸었다.
Gard 1/1	Pub. 22 - 39/52
Authors s	Gorzhevskiy, D. I.
Title e	The geological history of Rudniy Altay in the Paleozic era
	장치 많이 가능하는 생물이 되어 이동이 되는데 나는데 이 나는 사람들이 가능했다.
Periodical :	Dok. AN SSSR 101/4,731-733, Apr 1, 1955
Abstreat :	The basic traits are presented on the geological structure and geological history of the area known as Rudniy Altay formed during the middle Paleozoic era. Six USSR references (1936-1953).
Institution :	
Fresented by :	Academician N. S. Shatskiy, December 14, 1954
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